



Robert Millikan (top center) on the steps of Ryerson Laboratory, U. of Chicago, 1908. Other colleagues (L-R): A. A. Michelson, Carl Kinsey, Henry G. Gale

ROBERT A. MILLIKAN

Oil Drop Experiment Notebooks

NOTEBOOK TWO:
March-April 1912

PART 3 OF 3
From page 60 to page 84

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Abstract

Robert A. Millikan (1868-1953) began his experiments to measure the charge on the electron, e , in 1907. The experiments were performed in Ryerson Laboratory at the University of Chicago, where Millikan was professor of physics. For this work, and for work on the photoelectric effect, Millikan was awarded the Nobel Prize in physics in 1923.

Millikan gives his own account of the electron charge determination in his published autobiography in the chapter titled "My Oil-Drop Venture (e)" (Robert A. Millikan, *The Autobiography of Robert A. Millikan*, New York, 1950). With the aid of graduate students Louis Begeman, Harvey Fletcher, and J. Y. Lee, Millikan devised the method of measuring the rate of fall of a single electrically charged oil drop under the forces of gravity and electricity. From 1909 until the spring of 1912, Millikan reports, he spent every available moment in the laboratory on his oil-drop experiment. His first comprehensive, though to some extent preliminary, results were published in September 1910 in the journal *Science* as "The Isolation of an Ion, a Precision Measurement of Its Charge, and the Correction of Stokes' Law," *Science* 32: 436-448. He soon became embroiled in a controversy with the Viennese physicist Felix Ehrenhaft, who claimed to have found much smaller electric charges. Millikan went back to work on a new

set of experiments. By the spring of 1912 he had collected the data for what he termed “the final, absolute determination of the numerical value of the electron” (*Autobiography*, p. 84). Results were published in August 1913 in “On the Elementary Electrical Charge and the Avogadro Constant,” *Physical Review* 2: 109-43. This last, definitive set of experiments were recorded in the only two lab notebooks which Millikan preserved among his papers. These two notebooks are presented here in facsimile. They cover the period from October 1911 through April 1912 and contain what Millikan himself considered his conclusive, historic work on this problem.

For an analysis of Millikan’s notebooks and a defense of his experimental method, see the article by David Goodstein, “In Defense of Robert Andrews Millikan,” published in *American Scientist* 89/1 (Jan-Feb. 2001): 54.
<http://www.americanscientist.org/issues/num2/2001/1/in-defense-of-robert-andrews-millikan/1>

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Preferred citation

Robert A. Millikan Oil Drop Experiment Notebooks. Lab Notes Online. California Institute of Technology Archives. Retrieved [supply date of retrieval] from the World Wide Web:
http://resolver.caltech.edu/CaltechLN:LN_Millikan_R_2

Contact information

Archives, California Institute of Technology
Mail Code 015A-74
Pasadena, CA 91125
Phone: (626) 395-2704 Fax: (626) 793-8756
Email: archives@caltech.edu

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(51) Saturday Apr. 13, 1912 { start, $\theta = 22.95^\circ$ $b = \frac{94.60}{75.87}$
 Fourt Obs. 9:20 P.M. } End, $\theta = 23.0^\circ$ 5:58 P.M.

G	A	
28.456	32.2-63.2	62.988
14.3-28.4		
28.562	31.6-62.8	62.634
14.2-28.5		
28.468	31.4-62.8	
14.2-28.6	20.6-41.0	40.734
28.560		
-28.4	62.6	62.308
28.610	31.0-62.2	62.870
-28.7		
28.634	12.2-24.3	24.480
14.3-28.8	15.2-36.4	30.424
28.624		
14.2-28.6		
28.628		
8) 45.42		
28.57		
3		
28.60		
$\frac{1}{28.60} = .03497 \times 1021 =$		
$V_1 = .035704$		
$\log = -2.55275$		
$\frac{1}{2} = -1.2764$		

$\frac{1}{62.90} = .01590$
 $\frac{1}{40.80} = .02451$
 $\frac{1}{62.7} = .01595$
 $\frac{1}{30.45} = .03284$
 $V_1 + V_2 = .008473 \times 1021 = .008651$
 $\log = -3.93705$
 -1.27638
 3.1983
 6.41173
 3.70194
 -10.10983
 $c_1 = 5.126$
 $a_1 = 5.720$
 -16.4270
 3.7019
 -2.5528
 -14.6817
 -3.9370
 $3) -12.7447$
 -4.2482
 1.8601
 -2.7283
 1.8717
 $.05326 = \frac{1}{a}$
 $.05390 = \frac{1}{a}$
 74.42
 $74.77 = \frac{1}{h}$
 $832.5 + 13.8$
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 $817.5 + 15.1$
 $837.0 + 13.0$
 $838.0 + 12.9$
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 $830.0 + 14.2$
 $834.0 + 14.0$
 $820.5 + 14.8$
 $814.5 + 15.3$
 $827.5 + 14.4$
 $830.0 + 14.2$
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 $6) 5087$
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 $64.05 = e^8$
 $64.00 = e^8$
 $3\% \text{ low}$
 $1 + A \frac{L}{a} = 1.0452$
 $\log = .0192$
 2.0576
 00.53
 7.0963
 02.55
 6510
 $e = 4.776$

57) Saturday, Apr. 13, 1912
Seventh obs. at.

$\theta = 23.12$ $\phi = 94.89$
V. obs. at 11:35 AM. 76.42

G		F
40,518		17.760
40,522	32.0 - 63.2	63.100
40,382	96.4 - 174.4	114.2
40,542		15.138
40,650	14.8	27.716
40,696	14.0 - 28.0	15.060
40,478		38.606
40,458	38.4	38.382
40,494	38.4	172.626
40,802	86.4 - 172.2	
40,542		
40,54		
40,60		

$827.0 + 14.4$
 $825.0 + 14.1$
 $818.0 + 14.9$
 $813.0 + 14.9$
 $814.5 + 15.3$
 $826.5 + 14.5$
 $492.4 + 89.0$
 5013.0
 $826.5 + 14.5$
 $825.0 + 14.1$
 $817.0 + 15.0$
 $812.0 + 15.6$
 $809.0 + 16.0$
 $821.5 + 14.7$
 $491.0 + 90.4$
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58) Monday Apr. 15, 1912

 $\theta = 23.23$

$$\rho = \frac{6102}{5831} \frac{6441}{5531}$$

$$3.12 \quad 3.10$$
First Observation at
5:00 PM
 Volts at 5:00 PM 840.01
 12.7
 852.7

G	F	G	F
10.6 (2)		52.4	{ 21.6 } $\frac{1}{4}$
15.4 (4)			{ 41.8 }
12.0 (6)		51.6	{ 20.6 } $\frac{1}{4}$
12.8 (8)			{ 41.4 }
51.4	14.8		
13.0 (2)		51.32	
12.4 (4)		$\frac{1}{51.32} = 0.01944$	
13.6 (6)			
12.2 (8)			
51.0	14.6		
13.0 (2)			
12.6 (4)			
12.0 (6)			
13.0 (8)			
50.6	21.6		
12.4 (2)			
12.8 (4)			
12.6 (6)			
12.6 (8)			
50.4	21.6		
13.0 (2)			
13.0 (4)			
12.0 (6)			
13.4 (8)			
51.4	21.9		
13.2 (2)			
12.6 (4)			
12.8 (6)			
13.0 (8)			
51.6	116.0 (2)		
13.0 (2)	117.2 (4)		
12.6 (4)	117.5 (6)		
13.0 (8)	117.5 (8)		
51.6	110.3 (8)		
13.0 (2)	460.7		
12.6 (4)	110.0 fully div.		
52.4	{ 21.4 for distance }		
	{ 41.9 for whole dist }		
50.4	{ 21.0 = $\frac{1}{2}$ }		
	{ 41.8 = 1 }		
	14.8		

Diffs

$$\begin{array}{r} 0.2175 \\ 0.2185 \\ 0.2194 \\ 31.6554 \\ 0.2185 \end{array}$$

 at 5:41 PM
 Volts = 837.5
 13.0
 850.5

$$\begin{array}{r} 0.1944 \quad 0.1944 \\ 0.0217 \quad 0.0217 \\ 0.02161 \quad 0.02168 \\ 0.02168 \end{array}$$

$$\begin{array}{r} 1944 \quad 1944 \\ 4604 \quad 4604 \\ 31.06548 \quad 31.06548 \\ 0.02183 \quad 0.02183 \end{array}$$

$$\begin{array}{l} v_1 = 0.01985 \\ \log = -2.2978 \\ \frac{1}{2} v_1 = -2.1489 \end{array}$$

$$\begin{array}{l} v_2 = 0.02174 \times 1.31 \\ = 0.02220 \end{array}$$

$$\begin{array}{l} \log = -2.3464 \\ \frac{1}{2} v_2 = -1.1469 \end{array}$$

$$\begin{array}{l} 3.1983 \\ -6.6936 \end{array}$$

$$\begin{array}{r} 2.9296 \\ 9.7640 \end{array}$$

$$\begin{array}{r} 58.07 \\ 6 \end{array}$$

$$\begin{array}{r} e_1 = 58.13 \\ e_2 = 58.06 \end{array}$$

$$\begin{array}{r} 317.5288 \\ -6.5096 \end{array}$$

$$\begin{array}{r} 3234.02 \\ 8 \end{array}$$

$$\begin{array}{r} 16.4209 \\ 2.9296 \\ -2.2978 \end{array}$$

$$\begin{array}{r} -5.6483 \\ 2.3464 \\ -13.3019 \end{array}$$

$$\begin{array}{r} -5.7673 \\ 3.9296 \\ -4.2601 \end{array}$$

$$\begin{array}{r} 3.7398 \\ 5443 \text{ - line } \\ 5488 \text{ - } \frac{1}{2} \text{ } \end{array}$$

no 70

60/ Monday Apr. 15, 1914
Third Observation
at 8:55 P.M.

$$\theta = 23.05$$

$$p = \frac{62.21}{57.38} = 1.084$$

Vacts at 8:45

$$\begin{array}{r} 824.5 \\ 14.6 \\ 832.0 \\ 13.8 \\ \hline 1684.9 \end{array}$$

G		F
17.236	8.6 20.5	20.118
16.990	10.6 - 20.5	20.346
16.992		20.298
16.972		20.386
17.012		25.104
16.912	25.0	25.256
17.130	25.6-50.4	50.384
17.184	25.6-50.2	50.152
17.048	25.3-50.0	50.620
16.924	12.6-25.4	25.368
16.994	97.6-194.8	194.8
17.062	33.9-66.9	66.662
17.030	66.9	66.654
17.022	40.6	40.662
17.022	40.2	40.130
		50.2 (2)
		49.1 (4)
		48.4 (6)
		48.6 (8)

$$\frac{1}{20.39} = .04926$$

$$\frac{1}{25.20} = .03968$$

$$\frac{1}{50.48} = .01981$$

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$$\frac{1}{50.48} = .01981$$

$$\begin{array}{r} .05866 \\ .01492 \\ .07858 \\ .004911 \\ \hline .15707 \end{array}$$

$$\begin{array}{r} .5866 \\ .1492 \\ .7858 \\ .04911 \\ \hline .8847 \end{array}$$

$$\begin{array}{r} .5866 \\ .1492 \\ .7858 \\ .04911 \\ \hline .8847 \end{array}$$

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$$\begin{array}{r} .5866 \\ .1492 \\ .7858 \\ .04911 \\ \hline .8847 \end{array}$$

$$\begin{array}{r} .5866 \\ .1492 \\ .7858 \\ .04911 \\ \hline .8847 \end{array}$$

$$\frac{1}{17.053} = .05866 \times 10.21$$

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$$\frac{1}{17.053} = .05866 \times 10.21$$

$$\frac{1}{17.053} = .05866 \times 10.21$$

$$\begin{array}{l} \log = -2.7774 \\ \log = -1.3889 \end{array}$$

Beauty to show agreement between
the two methods of getting $v_1 + v_2$
Publication

No 63

Monday Apr. 15, 1912

$\theta = 23.06$

$$p = \frac{5249}{5.44}$$

Fourth Observation.
at 10:00 P.M.

Volts at 9:45

$$\begin{array}{r} 824.0 \\ 832.0 \\ 14.6 \\ 13.8 \\ \hline 1684.4 \end{array}$$

G	F
43.534 (6.4)	16.670 (6.4)
43.512 (6.4)	16.594 (6.6)
44.010 (6.4)	20.566 (6.4)
$3 \overline{) 131056}$ 43685 64	
43.75	

$$\frac{1}{1663} = .00013$$

$$\frac{1}{20.57} = .04861$$

$$\begin{array}{r} .02286 \\ .06013 \\ \hline 7108299 \\ 01185 \end{array}$$

$$\begin{array}{r} 2286 \\ 4861 \\ \hline 617147 \\ 01191 \end{array}$$

$$\begin{array}{r} 1191 \\ 1185 \\ 1152 \\ \hline 3528 \\ 1176 \end{array}$$

$$V_1 + V_2 = 0.1185 \times 10 \times 1 = .01200$$

$$\log = -2.0798$$

$$\begin{array}{r} -1.1841 \\ -3.1978 \\ -6.4617 \\ 3.2264 \\ \hline -9.2347 \end{array}$$

$$e_1 = 17.27$$

$$\begin{array}{r} -9.2348 \\ 3) -18.4694 \\ \hline -6.1565 \end{array}$$

$$\frac{1}{43.75} = .02286 \times 10 \times 1$$

$$= V_1 = .02334$$

$$\log = -2.3681$$

$$\frac{1}{2.11} = -1.1841$$

$$\begin{array}{r} -16.4289 \\ 3.2264 \\ -2.3681 \\ -14.0154 \\ -2.0792 \\ \hline 3) -13.9362 \\ -5.9787 \\ 7356 \\ -4.7148 \\ \hline 3.2859 \end{array}$$

$$1931 = \frac{1}{\mu}$$

ak

$$\begin{array}{r} -10.46136 \\ 3.2264 \\ -7.68776 \\ -2.0792 \\ \hline -5.60856 \\ 5.6787 \\ \hline -1.62986 \end{array}$$

$$\log \frac{1}{K} = .37014$$

$$\frac{1}{K} = 2345$$

$$1 - \frac{1}{K} = 1.345 = \frac{1}{\mu}$$

$$\log = 12.67$$

$$\begin{array}{r} 3.2264 \\ -9.8430 \\ \hline 12.67 \end{array}$$

$$b = .0006969$$

$$A = 9705$$

any 21st

(W) Tuesday, Apr. 16, 1917
First Observation at
5:27

$\theta = 20.00$
Volts at 4:24 P.M. $\rho = 6.56$

G	F
31.0	43.0
30.8	60.0
31.282	60.1
31.272	59.932
31.210	42.494
31.194	42.006
31.438	43.300
31.194	59.686
31.214	59.176
31.286	59.070
20.90	26.460
31.261	26.546
31.292	22.296

$\frac{1}{43.0} = .02326$
 $\frac{1}{60.0} = .01667$
 $\frac{1}{60.1} = .01664$
 $\frac{1}{59.932} = .01669$
 $\frac{1}{42.494} = .02353$
 $\frac{1}{42.006} = .02381$
 $\frac{1}{43.300} = .02310$
 $\frac{1}{59.686} = .01674$
 $\frac{1}{59.176} = .01690$
 $\frac{1}{59.070} = .01693$
 $\frac{1}{26.460} = .03779$
 $\frac{1}{26.546} = .03767$
 $\frac{1}{22.296} = .04484$

$\frac{836}{836} = 1$
 $\frac{13.1}{13.1} = 1$
 $\frac{1648.2}{1648.2} = 1$
 $\frac{830}{835} = .994$
 $\frac{142}{132} = 1.076$
 $\frac{1692.4}{1692.4} = 1$

Three puffs were made since last night's reading and this observation was made after the second puff - Reading now is 6.37 6.96 Taking 1/2 of the diff - add to last night's reading gives 6.56 cm.

$\frac{.03196}{.01664} = 1.921$
 $\frac{.03196}{.023634} = 1.353$
 $\frac{.01664}{.006943} = 2.398$
 $\frac{.023634}{.006949} = 3.402$

$\frac{3146}{264} = 11.917$
 $\frac{51034847}{6968} = 7319.8$
 $\frac{3196}{3770} = .848$
 $\frac{101.06966}{.006966} = 14534$
 $\frac{51296}{1959} = 26.18$

$v_1 v_2 = .006966 \times 1021 = .007106$
 $\log = -3.8517$
 -1.2568
 2.1978
 -6.3063
 3.2288
 -9.0775
 $e = 11.95$
 $e = 18.79$
 $e = 23$
 -9.0775
 $3) -18.1438$
 -6.0479
 $111.7 = e^3$

$a = -16.4209$
 $l = -4.8547$
 2.2788
 -3.5136
 -4.0378
 -4.1039
 -3.8517
 -1.9339
 $3) -13.5119$
 -4.5039
 9.169
 -4.9288
 $+3.0792 = 124.75 = \frac{1}{\rho}$
 $11.85 \text{ cm for length}$

$\log = -2.5136$
 $\frac{1}{2.1} = -1.2568$
 $\log = -2.5136$
 $\frac{1}{2.1} = -1.2568$

$A = \frac{2.28}{.006966} = 328.7$
 $B = \frac{2.28}{.006966} = 328.7$

Very thankful for multiple relations & very convincing. All times are corrected for chron.

Tuesday Apr. 16, 1912 $\theta = 20.30$ $\rho = \frac{6143}{5620} = 1.093$
 Third Observation at 8:45 Valt at 8:45 $830.0 + 14.2 = 844.2$

G	F	fit	Volts at 9:25
50.842	31.358	$\frac{1}{31.36} = .03189$	827.5 + 14.5 = 841.9
51.164	49.920	$\frac{1}{49.92} = .01985$	
51.776	50.662	$\frac{1}{50.66} = .01974$	
52.276	16.934	$\frac{1}{16.93} = .05900$	
52.276	16.962	$\frac{1}{16.96} = .05900$	
50.890	52.516	$\frac{1}{52.52} = .01902$	
51.682	51.542	$\frac{1}{51.54} = .01940$	
618.630			
5144			

Balanced
 Speed 0.0

$\frac{1}{5144} = .01944$
 $\text{Corr} = .01940 \times 1021 = .01961 = V$
 $\log = -2.2969$
 $\frac{1}{2} = -1.1484$

9:23

a
 -16.4109
 -2.9253
 -2.2969
 -15.6431
 -2.2986
 $3/13.3445$
 -5.7815
 5.992
 -4.2907
 3.7093
 l
 -4.8547
 5092
 -43455
 -5.7815
 5.640
 $3.664 = \frac{1}{a}$
 $3.620 = \frac{1}{a}$
 5120
 5070
 5070

$\log = -2.2986$
 1.1484
 3.1963
 -6.6483
 2.9253
 -9.7200
 5249
 67
 $e_1 = 51.88$

e_2
 $1-19.7151$
 $3-17.4302$
 -6.4767
 $299.7 \times 10^{-5} = e_2$

a
 70.46136
 2.9253
 -7.39196
 -3.1834
 -4.20856
 $3/13.3445$
 -5.7815
 5.992
 -4.2907
 3.7093

a
 70.46136
 2.9253
 -7.39196
 -3.1834
 -4.20856
 $3/13.3445$
 -5.7815
 5.992
 -4.2907
 3.7093

911
 80
 3.884
 5911
 7730

No log

- Dr CV Raman Indian Assn for the Cultivation of Science Calcutta India --NW
- Prof The Svedberg Nobel Institute. Stockholm ~~Sweden~~ ^{Sweden} ~~Sweden~~ --NW
- Dr N Bohr 35 Gersonvej Hellerup Copenhagen ~~Denmark~~ ^{Denmark} --NW
- Dr R Sievert. Nobel Institution Stockholm Sweden. (..NW)
- Prof. Satyendra ~~Ray~~ Ray. Canning College Lucknow India --NW
 9, Linden Hill 1 Highland House Rd 117 Finsbury Pavement London
- Professor A. Majorana Politechni School Turin Italy --NWAB
- Dr C.G. Darwin Christ College Cambridge Eng --NWAB
- Prof E.C.C. Baly University of Liverpool. Eng --NWAB
- " Frank Horton University of London --NWAB
- Dr F.W. Aston Univ of Cambridge Cambridge Eng --NWAB
- Prof R Whiddington D.Sc. University of Leeds Leeds Eng
- Dr L. Silberstein Eastman Kodak Co. Rochester N.Y.
- A. Baeuillier 29 Rue chateaubriand Paris France
- Prof Dr Jc Stöckl Regensburg Aegy d 21 Germany
- Dr. Dr. Giorgio Abetti Osservatorio Grechi-Firanci Florence Italy

Books on Electromagnetism

J. J. Conly

R. W. Wood

G. B. Millikan

Ch. Fabry

H. Abraham

C. E. Mandershall

Carl Smyder

H. Walker

A. A. Michelson

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M. L. de Broglie ... NW 2

29 rue Chateaubriand

... de Bruché

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... Dr. Gustaf. Cronquist

Swedish Royal Acad of Sci

Stockholm Sweden

... Dr. N. Bohr ... NW

46 Astr. Søgade

Copenhagen. Denmark

... Prof. J. H. Jeans ... NW

8 Cromwell Gate

Cheltenham SW London

... Dr. H. T. Osborn Am. Mus. Nat. Hist

Dr. M. E. Hille ... NW

Russia. Sweden etc

Prof A Joffe Physch. Laboratorium d. Polytechnicum St Petersburg Russia 20 ☒ X

Dr L. Millen Öholm. Nobelinstitut Stockholm Sweden ☒ X = NW

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" P. Zeeman " " ☒ X = NW

" H. Kamerlingh Onnes. Leiden Holland. Univ. of Leiden. ☒ X = NW

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- Prof R A Sampson Royal Observatory Edinburgh VI 0 0 =
- " T Svedby The University Glasgow VI 0 0 X = NW
- ~~Prof W C Swanwick~~ ^{FRS} ~~Cornegie Institution~~ ^{Dept Physical} ~~Washington D.C.~~ ^{magnetics} VI 0 0 X
- ~~S P Thomson FRS~~ ~~Moreland~~ ~~Chislet Road~~ ~~West Hampstead~~ ~~1933~~ VI 0 0 X
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- ~~Mr S B M. Lomon~~ ~~Univ of Birmingham~~ ~~England~~ 0
- " E G Darrin ~~Cambridge Lab~~ ~~to Sir G P~~ 0
- Professor A Gray FRS Univ of Glasgow 0 X = NW
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- Prof W Le féval College Municipal Secondary School for Boys New North Road NW

- Prof Gilbert A. Lewis. Univ. of California Berkeley Cal $\odot \square X = \sim NWAB$
- Prof R C M ^{agassiz} ~~an~~ Mass. Inst of Technology Boston Mass $\odot \square X =$
- Prof E B Ineson " " " " $\odot \square X = \sim NWAB$
- " H N Davis Harvard Univ. Cambridge Mass $\odot \square X = \sim NWAB$
- " R C Tolman. Chemical Warfare Service. Washington DC $\sim NW$ $\sim NWAB$
- " Theo. Lyman Harvard Univ. Cambridge Mass $\odot \square X = \sim NWAB$
- " P W Bridgman " " " " $\odot \square X = \sim NWAB$
- " ~~Wm Duane~~ ~~G. N. Pierce~~ " " " " $\odot \square X = \sim NWAB$
- " ~~Eleuth. Thomson~~ G E Co. Lynn Mass $\odot \square X = \sim NWAB$
- " A W Duff. Worcester Polytechnic Inst Worcester Mass $\odot \square X = \sim NWAB$
- " A G. Foster Clark Univ. Worcester Mass $\odot \square X = \sim NWAB$
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- " John Zelensky " " ~~Yale Univ. New Haven Conn~~ $\sim NWAB$ $\odot \square X = \sim NWAB$
- " O M Stewart " " ~~Yale Univ. New Haven Conn~~ $\sim NWAB$ $\odot \square X = \sim NWAB$
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- " W F Hildebrand Bureau of Standards Wash. D C $\odot \square X =$
- " W. F. G. Swann University of Minnesota Minneapolis Minn. $\sim NWAB$ $\odot \square X =$

Determination of A & K

$$\text{Information } X = 6.4 \text{ MeV } (1 + A_2^2)$$

$$K = \frac{e^2 - e_0^2}{\frac{1}{A} e_0^2} \text{ or } A = \frac{e^2 - e_0^2}{\frac{1}{A} e_0^2}$$

$$K = \frac{79.0 - 6.14}{467.5 \times 6.14} = \frac{173.5}{2852.6} = .006251$$

where $p = 0.44 \times 23 = 10.12$ $e = \text{cm}$

Since $A = \frac{K}{q} \log K = -4.7960$

" " " " $K = -4.8547$

" " " " $K = -1.9413$

where $A = .8736$

- Dr Silberman Research Dept Eastman K Co Rochester NY $\sim NWAB$
- " R W. Wood Johns Hopkins Baltimore $\odot X =$
- " ~~Wm Duane~~ ~~Stanford Pa~~ $\odot X =$
- " R K Mc Clung Univ of Manitoba Winnipeg $\odot X =$
- " H A Wilson Rice Inst. Houston Texas $\odot X =$
- " E W Mason ~~Urbana Ill~~ $\odot \square X =$
- " G. H. Hall. Harvard Univ. Cambridge Mass $\odot \square X =$
- " ~~Wm P. Phipps~~ ~~Columbia~~ ~~New York City~~ $\odot X =$
- " ~~C. E. Ruck~~ ~~Harvard Univ. Cambridge Mass~~ $\odot X =$
- " ~~P. L. Webster~~ ~~Smithsonian Institution~~ $\odot X =$
- " ~~Wm. A. Ruggan~~ ~~1945~~ ~~Stanford Pa~~ $\odot X =$
- " Prof T. W. Richards. Harvard Univ. Cambridge Mass $\odot X =$
- " Otto Stuehlermann Univ of Penn Philadelphia Pa $\sim NWAB$
- " K T Compton Princeton N J $\sim NWAB$
- " ~~T. R. Mendenhall~~ ~~Wisc.~~ $\sim NWAB$

Final value 1c

$$C_0^* = C_1^* - d \text{ term} = \left(C_1^* - d \frac{79.00 - 6.115}{4675} \right) = \left(C_1^* - d \times 0.03620 \right) \times 10^{-2}$$

1	d=8856	$C_1^* = 6440$	$C_2^* = 5.168$	$2040 \times 1.0472 = 2136.384$	$6440 - 3396 = 3044$	$C_1^* = 6.101$
2	d=5491	$C_1^* = 6308$	$C_2^* = 5.010$	$2040 \times 1.0472 = 2136.384$	$6308 - 3396 = 2912$	$C_1^* = 6.101$
3	d=5991	6335	5.041	$2040 \times 1.0472 = 2136.384$	$6335 - 3396 = 2939$	$C_1^* = 6.106$
4	d=7464	6400	5.120	$2040 \times 1.0472 = 2136.384$	$6400 - 3396 = 3004$	$C_1^* = 6.115$
5	d=4790	6293	4.991	$2040 \times 1.0472 = 2136.384$	$6293 - 3396 = 2897$	$C_1^* = 6.110$
6	d=7140	6382	5.098	$2040 \times 1.0472 = 2136.384$	$6382 - 3396 = 2986$	$C_1^* = 6.109$
7	d=8923	6459	5.190	$2040 \times 1.0472 = 2136.384$	$6459 - 3396 = 3063$	$C_1^* = 6.115$
8	d=6068	6357	5.069	$2040 \times 1.0472 = 2136.384$	$6357 - 3396 = 2961$	$C_1^* = 6.125$
9	d=5604	6324	5.028	$2040 \times 1.0472 = 2136.384$	$6324 - 3396 = 2928$	$C_1^* = 6.109$
10	d=6092	6333	5.040	$2040 \times 1.0472 = 2136.384$	$6333 - 3396 = 2937$	$C_1^* = 6.100$
11	d=4828	6282	4.981	$2040 \times 1.0472 = 2136.384$	$6282 - 3396 = 2886$	$C_1^* = 6.098$
12	d=6119	6354	5.065	$2040 \times 1.0472 = 2136.384$	$6354 - 3396 = 2958$	$C_1^* = 6.120$
13	d=8507	6436	5.166	$2040 \times 1.0472 = 2136.384$	$6436 - 3396 = 3040$	$C_1^* = 6.111$
14	d=5544	6320	5.012	$2040 \times 1.0472 = 2136.384$	$6320 - 3396 = 2924$	$C_1^* = 6.098$
15	d=4475	6275	4.971	$2040 \times 1.0472 = 2136.384$	$6275 - 3396 = 2879$	$C_1^* = 6.104$
16	d=6159	6345	5.056	$2040 \times 1.0472 = 2136.384$	$6345 - 3396 = 2949$	$C_1^* = 6.109$
17	d=4631	6262	4.979	$2040 \times 1.0472 = 2136.384$	$6262 - 3396 = 2866$	$C_1^* = 6.103$
18	d=4944	6313	5.017	$2040 \times 1.0472 = 2136.384$	$6313 - 3396 = 2917$	$C_1^* = 6.124$
19	d=4609	6300	5.001	$2040 \times 1.0472 = 2136.384$	$6300 - 3396 = 2904$	$C_1^* = 6.124$
20	d=5165	6313	5.016	$2040 \times 1.0472 = 2136.384$	$6313 - 3396 = 2917$	$C_1^* = 6.116$
21	d=8078	6282	4.982	$2040 \times 1.0472 = 2136.384$	$6282 - 3396 = 2886$	$C_1^* = 6.126$
22	d=7826	6434	5.145	$2040 \times 1.0472 = 2136.384$	$6434 - 3396 = 3038$	$C_1^* = 6.120$

$$C_0^* = 6.111$$

$$C_1^* = 4.777$$

American Physicists

Dr Irving Langmuir	Schenectady N.Y. G.E. Research Lab	..NWAB
Prof A Kennedy	Harvard Univ	OX = ..NWAB
Laboratory of Physical Chemistry (Prof Washburn) U of Ill		
Prof W B Pogram	Columbia New York City	OX = ..NWAB
" M J Papin	"	OX = ..NWAB
" W S Hallowell	"	"
" W S Helms	Emory University Emory Ga	OX = ..NWAB
" C C Muller	Case School Cleveland O	OX = ..NWAB
" A D Cole	Ohio State U Columbus O	OX = ..NWAB
" A W Goodspeed	Univ of Pa Philadelphia Pa	OX = ..NWAB
" W S Franklin	Lehigh Univ South Bethlehem Pa	OX = ..NWAB
" Carl Barus	Brown " Providence R.I.	OX = ..NWAB
" W J Humphreys	Maths Bureau Wash DC	OX = ..NWAB
" V C E Mendenhall	Univ of Wis Madison Wis	OX = ..NWAB
" V E Peraval Lewis	" " Cal Berkeley Cal	OX = ..NWAB
" V G E Hale	Mt Wilson Cal	OX = ..NWAB
" A S Eve	McGill Univ Montreal Canada	OX = ..NWAB
" H T Barnes	" " " "	OX = ..NWAB
" J C Mc Lennan	Toronto Univ Toronto Can	OX = ..NWAB
" Frank Allen	Univ of Manitoba Winnipeg Canada	OX = ..NWAB
" W G Cady	Wash Univ Middletown Conn	OX = ..NWAB
" B B Boltwood	New Haven Conn	OX = ..NWAB
" H A Bumstead	1701 Mass Ave Wash DC	OX = ..NWAB
" L A Bauer	Carnegie Institution Wash DC	OX = ..NWAB
" W W Coblenz	Bureau of Standards Wash DC	OX = ..NWAB
" A L Day	Geophysical Lab Wash DC	OX = ..NWAB
" E B Rosa	Bureau of Standards Wash DC	OX = ..NWAB
" S W Stratton	" " " "	OX = ..NWAB
Prof R S Woodward	Carnegie Institution Wash DC	OX = ..NWAB
Prof Henry Crow	Evansston Ill Northwestern Univ	OX = ..NWAB
" J F Hayford	" " " "	OX = ..NWAB
" A L Foley	Univ of Ind Bloomington Ind	OX = ..NWAB
" G W Stewart	" " Iowa Iowa City Iowa	OX = ..NWAB
" Jos S Ames	Johns Hopkins Baltimore Md	OX = ..NWAB
" A L Kimball	Amherst College Amherst Mass	OX = ..NWAB
" H M Goodwin	Mass Inst of Technology Boston	OX = ..NWAB
" D F Comstock	" " " "	OX = ..NWAB

Irving Langmuir Scholarship

A W Hull

Prof Richard Tolman U of Ill Urbana

Prof P Carman

Density of Clock Oil

Taken by R. Millikan April 5/12

Weight of empty bottle	10.9181
" " Bottle full	20.1586
" " " water	20.9338

Temperature of both oil & water
exactly 23°

" of oil alone = 9.2355
" " water " = 10.0157

Relative densities = 9.221

abs density, oil = 9.221 x 9.976 = 9.199

$\mu = 0.01125$
 $\frac{0.01125}{0.000125} = 90$

$$e = \frac{2}{3} \pi \left(\frac{r}{\lambda} \right)^2 \frac{1}{g(r-p)} \frac{A(V_1+V_2)V_1^2}{PD} = \frac{4\pi}{3} \left(\frac{9.1825}{2} \right)^2 \frac{1}{(980.3/919.00)} \frac{1}{2} \frac{1.6(V_1+V_2)V_1^2}{PD}$$

$$= \frac{\pi \times 64 \times (0.0082125)^2 \times 100}{(980.3 \times 91873)^2} \times \frac{(V_1+V_2)V_1^2}{PD \text{ water}}$$

Calc'n of the
constant coeff
of $(V_1+V_2)/V_1$
PD

$$= \frac{\pi \times 640 \times (0.0082125)^2}{(980.3 \times 9187)^2} \frac{(V_1+V_2)V_1^2}{PD \text{ water}} \quad \log(11299426) = 7.053$$

$$21 - 10.74242 = 10.25758$$

$$-5.371714$$

$\log \pi = .49715$
 $11.640 = 2.80618$
 $11.11^2 = 5.371714$
 -2.67504
 11.47724
 -3.19776

$\log 980.3 = 2.99136$
 $11.9167 = 1.96319$
 $21 - 2.95455 = 18.04545$
 1.47728

Calc'n of $\frac{3}{2} \pi$ $\frac{2}{3} \pi = \frac{2}{3} \pi$ $\frac{3}{2} \pi = \frac{3}{2} \pi$ $\frac{V_1+V_2}{V_1}$

$$e = \frac{2}{3} \pi \frac{A^2(r-p)^3}{F} \left(\frac{V_1+V_2}{V_1} \right) \therefore A^2 = \frac{3Fe}{4\pi g(r-p)(V_1+V_2)}$$

$$e = \frac{3Fe}{4\pi g A^2(r-p)} \frac{V_1 PD_{\text{water}}}{V_1+V_2} = \frac{4.377}{4\pi \times 980.3 \times 9195} \frac{V_1 PD}{V_1+V_2}$$

$\log e = -10.67916$
 6.25824
 -16.42091
 $1.46355 = 9195$
 1.636×10^{-16}
 6.25824

This is just 1/77th from the value found in calc'n
which means that it must be reduced 57%

shown 227 smaller
than number with him
4.11 15.768 x 10⁻³

Calc'n of ak $e = \frac{6\pi \eta a^2 \omega}{F} \frac{V_1(V_1+V_2)}{PD}$

$$ak = \frac{F}{6\pi \eta a^2 \omega} \frac{PD}{V_1(V_1+V_2)}$$

$$ak = \left(\frac{PD}{\eta a^2} \right) \frac{1}{30006 \pi \omega}$$

$\log ak = \log 10.46136 \times \frac{PD}{\eta a^2} \frac{1}{30006 \pi \omega}$

Calibration of Hipp Chronograph

30 sec interval	20 sec interval	10 sec interval	16 sec. Int.	Exphistic 20 sec.
14981	14945	9979	5003	8012
14994	14959	9996	4980	7980
14991	14960	9987	4992	8003
14993	14986	9962	5011	8013
14965	14986	9981	4995	8015
14982	14986	9999	4996	8005
15001	15003	10012	5014	8011
14970	14964	9983	4990	8001
14999	15009	9972	5010	8005
14965	14963	9983	5009	7996
9936	99492	9964	5009	.0041
14.9836	14.986	9912	50009	80041
29.972	29.972	998345	100018	160082
Corr = .1%	19.96690	Corr = .06%	Corr = 0	Corr = 0

6 sec	40 sec	60 sec	112 sec	
3001	99991	29969	111.864	- .13
3031	99985	29950		
3037	99984	29962	116. sec	
3008	99955	29939	115.722	-.24%
2995	99984	29948		
3017	99950	5 149768	120. sec	
3003	99960	299536	119.782	-.14%
2980	99954	599072		
2989	99996			
2988	99959			
3026	99718			+ .20%
3013	199718			
36588	399436			
300733				
60146	+ .15%			
325%				

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 Journal de Physique 42 Rue de Valenciennes Paris
 Fortschritts der Physike. Prof. Karl Scheel Melnarsdorf Gtingal Str 46 Berlin Gering

France

Paris France

- Prof. H. A. D. Anson Prof. Physique College de France 12 $\odot \odot X$
 Mrs. Dr. P. Curie Laboratoire de Radium Rue Pierre Curie Paris $\odot \odot X = \text{NW}$
 Ch. d. Guillaume Bureau International Sevres 12 $\odot \odot X = \text{NW}$
 M. Ch. d. Guye Gonin Univ. Zurich 12 $\odot \odot X = \text{NW}$
 " Svingedau Lille $\odot \odot$
 " Roche Nancy $\odot \odot$
 " Blanc Caen $\odot \odot$
 " H. Abraham Prof. de la Sorbonne Paris 12 $\odot \odot X = \text{NW}$
 " M. A. Cotton Ecole Normale Supérieure Paris 12 $\odot \odot X = \text{NW}$
 " Charles Fabry Univ. of Paris. Laboratoire de Physique 6 rue Victor Cousin Paris France $\odot \odot X = \text{NW}$
 Prof. A. Ricci Bologna Italy 12 $\odot \odot X =$
 M. L. Du Broglie 29 rue Chateaubrand Paris $\odot \odot X = \text{NW}$
 " H. Brissou Montellon $\odot \odot X = \text{NW}$
 " Ed. Becquerel Bureau du Journal de Physique 5 Rue de la Santé $\odot \odot = \text{NW}$
 Prof. Jean Perrin Sorbonne Paris France $\odot \odot X = \text{NW}$
 " L. Langevin College de France Paris $\odot \odot X = \text{NW}$
 " G. Lippmann University of Paris $\odot \odot X = \text{NW}$
 " M. Marcel Brillouin College de France Paris France $\odot \odot X = \text{NW}$

German Physicists

- Herrn Geheimrat Prof. Johannst. Bunsen Str. 1
 X Prof. Dr. P. Haber Stra. 14 Dahlen Berlin VI 0 X NW
 Prof. Max Planck Wangerheim Str. 21 Grunewald VI 0 X NW
 Prof. W. Kossel Phys. Inst. d. Tech. Hochschule Munchen Germany NW
 Prof. Dr. H. Rubens Physikalisches Institut Berlin VI 0 X NW
 Dr. O. v. Beyer Landwirtschaftliche Hochschule Berlin VI 0 X NW
 Dr. Robert Kohl Physikalisches Institut Berlin VI 0 X NW
 Dr. Peter Brunsheim Technische Hochschule Berlin VI 0 X NW
 Dr. Erich Regener Stuttgart Germany NW
 Dr. J. Frank Göttingen Physikalische Inst. NW
 Prof. E. Geicke Karlsruhe Institut Dahlen bei Berlin NW
 Dr. H. Hupfner Rensselaerstr. March Str. Charlottenburg Berlin I 0 X NW
 Prof. A. Schell Berlin-Dahlem NW
 Dr. A. Lueken Berlin-Dahlem NW
 Dr. R. Ladenburg Brissau Physikal. Institut der Univ. NW
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 Dr. G. B. H. van der Bijl Dept. Mines and Industries Union South Africa Pretoria S.A. NW
 Dr. H. Damber " " " NW
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